

PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification: B32B 31/00	A1	(11) International Publication Number: WO 95/00333
		(43) International Publication Date: 5 January 1995 (05.01.95)

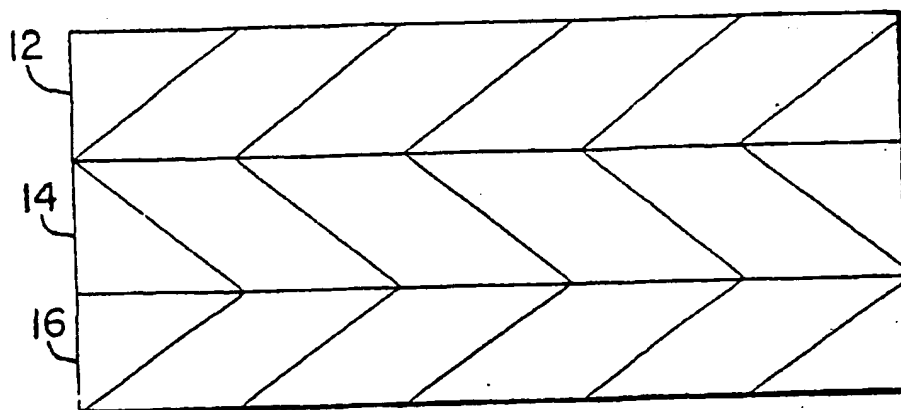
(21) International Application Number: PCT/US94/07059

(22) International Filing Date: 22 June 1994 (22.06.94)

(30) Priority Data: 08/082,226 24 June 1993 (24.06.93) US

(71) Applicant: AMERICAN NATIONAL CAN COMPANY  
[US/US]: 8770 West Bryn Mawr Avenue, Chicago, IL  
60631 (US).(72) Inventors: ECKSTEIN, John, P.; 1206 Glenayre Drive,  
Neenah, WI 54956 (US). ZHENG, Johnny, Q.; 2113  
Clover Lane, Appleton, WI 54915 (US). NORDNESS,  
Mark, E.; 951 Grove Street, Neenah, WI 54956 (US).  
LIND, Keith, D.; 1806 N. Edgewood, Appleton, WI 54911  
(US). WALBRUN, George, H.; 1047 Ida Street, Menasha,  
WI 54952 (US). SHEPARD, Mary, E.; 1036 Grove Street,  
Oshkosh, WI 54901 (US). JONES, Gregory, K.; 2653  
Sunnyview Road, Appleton, WI 54914 (US). SEEKE,  
Gregory, J.; 2901 Daniel Court, Oshkosh, WI 54904 (US).(74) Agent: O'ROURKE, Thomas, A.; Wyatt Gerber Burke &  
Badic, 5th floor, 645 Madison Avenue, New York, NY  
10022 (US).(81) Designated States: AU, CA, JP, NZ, European patent (AT, BE,  
CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT,  
SE).Published  
With international search report.

(54) Title: IMPROVED STRUCTURES OF POLYMERS MADE FROM SINGLE SITE CATALYSTS



(57) Abstract

Copolymers of ethylene and alpha olefins that have been formed by a polymerization reaction in the presence of a single site catalyst, such as a metallocene, are used as a film or as a layer in multiple layer films, including molecularly oriented and irradiated heat shrinkable films (12, 14, 16). Novel blends of the copolymers with other polymeric materials are disclosed and used as a film or a layer in a film particularly in molecularly oriented and heat shrinkable films. Bags made from the multiple layer films are especially useful for shrink packaging primal cuts of meat. Processes for the formation of flexible films and packages made therefrom are also disclosed.

*FOR THE PURPOSES OF INFORMATION ONLY*

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
AU	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Kenya	RO	Romania
CA	Canada	KG	Kyrgyzstan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic of Korea	SD	Sudan
CG	Congo	KR	Republic of Korea	SE	Sweden
CH	Switzerland	KZ	Kazakhstan	SI	Slovenia
CI	Côte d'Ivoire	LI	Liechtenstein	SK	Slovakia
CM	Cameroon	LK	Sri Lanka	SN	Senegal
CN	China	LU	Luxembourg	TD	Chad
CS	Czechoslovakia	LV	Latvia	TG	Togo
CZ	Czech Republic	MC	Monaco	TJ	Tajikistan
DE	Germany	MD	Republic of Moldova	TT	Trinidad and Tobago
DK	Denmark	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	US	United States of America
FI	Finland	MN	Mongolia	UZ	Uzbekistan
FR	France			VN	Viet Nam
GA	Gabon				

IMPROVED STRUCTURES OF POLYMERS MADE  
FROM SINGLE SITE CATALYSTS

BACKGROUND OF THE INVENTION

Polymeric materials have many applications in packaging structures. They are used as films, sheets, lidstock, pouches, tubes and bags. These polymeric materials may be employed as a single layer or one or more layers in a structure. Unfortunately, there are countless polymeric materials available. Furthermore, resin suppliers frequently have a tendency to claim many more applications for a product than the product is actually suitable for. In addition, in view of the specialized applications and processing problems that are encountered despite the suppliers claims, one skilled in the art can not tell whether a particular resin will be suitable for an application unless tested. However, for various reasons there are frequently drawbacks to the use of many of these polymeric materials. For example, ethylene vinyl alcohol is an excellent oxygen barrier material for use in packaging food products. However, this polymeric material can be affected by moisture that is present in the atmosphere or the packaged product. As a result, it is frequently found that some polymeric materials are better for certain applications than others.

One area where there is a need for suitable resins in film applications is in the area of heat shrinkable films. Heat shrinkable polymeric films are commonly used in packaging meats, particularly primal meat

cuts and other large pieces of meat. While this description will detail the usage of films for packaging meat and meat by-products, it will be understood that these films are also suitable for packaging a myriad of other products, both including food products and non-food products.

Some of the films embodying the present invention are intended to be used by meat packers in the form of heat shrinkable bags with one opened end, which bags are closed and sealed after insertion of the meat. After the product is inserted, air is usually evacuated from the package and the open end of the bag is closed. Suitable methods of closing the bag include heat sealing, metal clips, adhesives etc. Heat is applied to the bag once sealing is completed to initiate shrinkage of the bag about the meat.

In subsequent processing of the meat, the bag may be opened and the meat removed for further cutting of the meat into user cuts, for example, for retail cuts or for institutional use.

Suitable shrink bags must satisfy a number of criteria. Many bag users seek a bag that is capable of surviving the physical process of filling, evacuating, sealing and heat shrinking. For example, during the shrinking process great stress can be placed on the film by the sharp edges of bone in the meat. The bag must also have sufficient strength to survive the material handling involved in moving the large cuts of meat, which may weigh

a hundred pounds or more, along the distribution system.

Because many food products including meat deteriorate in the presence of oxygen and/or water, it is desirable that the bags have a barrier to prevent the infusion of deleterious gases and/or the loss or addition of moisture.

Conventional packaging for many products has frequently been made of multiple layer films having at least three layers. These multiple layer films are usually provided with at least one core layer of either an oxygen barrier material such as a vinylidene chloride copolymer, ethylene vinyl alcohol, a nylon or a metal foil preferably aluminum. Heat shrinkable meat bags, for example, have generally used vinylidene chloride copolymers. The copolymer of the vinylidene chloride may, for example, be a copolymer with vinyl chloride or methyl acrylate. Collapsible dispensing tubes have generally used one or more foil layers. The foil layers in addition to supplying an oxygen barrier also provide the dispensing tube with "deadfold", i.e., the property of a collapsible dispensing tube when squeezed to remain in the squeezed position without bouncing back.

Outer layers of films used in packaging food products can be any suitable polymeric material such as linear low density polyethylene, low density polyethylene, ionomers including sodium and zinc ionomers such ionomers include Surlyn, ethylene vinyl acetate etc. In conventional shrink bags, the outer layers are generally

linear low density polyethylene or blends thereof. Suitable outer layers for meat bags are taught by U.S. Patent No. 4,457,960 to Newsome, the disclosures of which are incorporated herein by reference.

While conventional films have been suitable for many applications, it has been found that there is a need for films that are stronger and more easily processed than conventional films. In meat bags, there is a need for films and bags that have superior toughness and sealability and the ability to undergo cross-linking without undue deterioration. Thus, it is an object of the present invention to provide improved structures, including single and multi-layer films, sheets, lidstock pouches, tubes and bags. In particular, structures for use in shrink bags wherein the shrink bags are capable of withstanding production stresses and the shrink process.

#### SUMMARY OF THE INVENTION

The structures of the present invention may be single or multilayer films, sheets, lidstock, pouches, containers, tubes and bags where at least one layer contains a polymer usually a copolymer formed by a polymerization reaction in the presence of a single site catalyst such as a metallocene. Examples of such a polymer are ethylene and propylene polymers and copolymers thereof. One preferred copolymer is a copolymer of ethylene and an alpha olefin where such alpha olefin has a carbon chain length of from  $C_3$ - $C_{20}$ . The structures of the present invention may also include blends of polymers and

copolymers formed by a polymerization reaction with a single site catalyst or blends of a polymer and copolymer formed by a polymerization reaction with a single site catalyst and another polymeric material. Examples of suitable polymers for blending include: high and medium density polyethylene (HDPE, MDPE), linear low density polyethylene (LLDPE), low density polyethylene (LDPE), ethylene vinyl acetate (EVA), ultra low density polyethylene (ULDPE or VLDPE), and ionomers such as Surlyn.

The present invention may also be a multilayer structure of at least three layers wherein the core layer is a barrier layer. In one embodiment of the present invention, there may be a first outer layer of an ethylene or propylene polymer or copolymer formed by a polymerization reaction in the presence of a single cell catalyst, a barrier layer and a second outer layer of a polymeric material. The second outer layer may be an ethylene or propylene polymer or copolymer formed by a polymerization reaction in the presence of a single site catalyst or a layer of another polymeric material such as high density polyethylene, medium density polyethylene, linear low density polyethylene, ultra low density polyethylene, low density polyethylene, ethylene vinyl acetate, an ionomer or blends thereof. The first outer layer may also be a blend of the ethylene copolymer with another suitable polymeric material such as described above. A preferred polymer formed by a single site

catalyst is a copolymer of ethylene and an alpha olefin such as octene-1. Additional layers such as adhesive layers or other polymeric layers may be interposed in the structure between one or both of the outer layers or on top of one or both of the outer layers. The structure of the present invention may be rendered oriented either uniaxially or biaxially and cross-linked by any suitable means, such as for example irradiation or chemical cross-linking.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a side view of a three layer structure of the present invention.

Figure 2 is a side view of a five layer film of the present invention.

Figures 3-6 are examples of the structure of metallocene catalysts used in the polymerization of the polymer used in the structures of the present invention.

#### DETAILED DESCRIPTION OF THE INVENTION

The structures of the present invention include films, sheets, lidstock, pouches, containers, tubes and bags. These structures may be a single layer or multi-layer structure. The structures are comprised of polymers that have been polymerized in the presence of a single site catalyst, such as a metallocene. A metallocene is a complex organometallic molecule typically containing zirconium or titanium, with a pair of cyclic alkyl molecules. More specifically, metallocene catalysts are usually compounds with two cyclopentadiene rings fixed to



the metal. These catalysts are frequently used with aluminoxanes as a co-catalyst or an activator, one suitable aluminoxane is a methylalumoxane (MAO). Besides, titanium and zirconium, hafnium may also be used as the metal to which the cyclopentadiene is bonded. Alternative metallocenes may include Group IVA, VA and VIA transition metals with two cyclopentadiene rings. Also mono-cyclopentadiene rings or silyl amides may alternatively be in the metallocene instead of two cyclopentadienes. Other metals to which the cyclopentadiene may be attached may include the metals in the lanthanide series. Figures 3, 4, 5 and 6 show representative metallocenes that are suitable single site catalysts.

While the reaction mechanism is not completely understood, it is believed that the metallocene, single site catalyst confines the copolymerization reaction to a single site over the polymer thus controlling comonomer placement and side chain length and branching. The copolymers formed from metallocene single site catalysts are highly stereo regular products with narrow molecular weight distribution. The metallocenes can be used to polymerize ethylene, propylene, ethylenic and acetylenic monomers, dienes and carbon monoxide. Comonomers with ethylene and propylene include styrene, substituted styrene, vinyl, acrylonitrile, methyl acrylate, methyl methacrylate and 1,4 - hexadiene. The metallocene single site catalysts are capable of producing isotactic polymers and syndiotactic polymers, i.e., polymers in which the

crystalline branches alternate regularly on both sides of the back bone of the polymer. There are two general types of single site catalyst reactions. The first are nonstereoselective catalysts reactions which have been developed by Exxon and Dow and which are used to make Exxon's Exact resins and Dow's CGCT resins. See Figs. 3 and 4. The second type of reactions are stereoselective catalysts developed by Hoechst and Fina for stereo specific polymerization particularly of polypropylene and other olefins such as butene-1, and 4 methylpentene-1. See, e.g., Figures 5 and 6.

The ethylene alpha olefins polymerized by a single site catalyst have low crystallinity and a density that ranges from 0.854 to 0.97 gm/cc. Although this density range is similar to conventional ethylene polymers, i.e., LDPE, LLDPE and ULDPE, the polymers in the structures of the present invention have a narrow molecular weight distribution and homogeneous branching. The molecular weight distribution of the preferred polymers may be represented by the formula

$$MWD = M_w/M_n = <2.5$$

In addition, the melt processability of these polymers ( $I_{10}/I_2$ ) has a range of about 5.5 to about 12 while conventional homogenous polymers are generally less than 6.5 at an MWD of 2. The melt tension of these polymers is in the range of about 1.5 to 3.5 grams.

The MWD of these polymers may be determined using a Waters' 150 GPC at 140°C with linear columns (103,

A-10<sup>6</sup> A<sup>0</sup>) from Polymer Labs and a differential refractometer detector. Comparison of the MWD of a 1MI, 0.920 density CGCT polymer with that of 1MI, 0.920 density conventional LLDPE illustrates the very narrow MWD of the CGCT polymers which usually have a  $M_w/M_n$  of approximately 2 compared to 3 or greater for LLDPE.

A preferred ethylene copolymer is a copolymer of ethylene and a C<sub>3</sub> to C<sub>20</sub> alpha olefin. A preferred copolymer is a low modulus ethylene octene copolymer sold by Dow. This copolymer is formed by Dow's constrained-geometry catalyst technology which uses a single site catalyst such as cyclo-pentadienyl titanium complexes. As best understood, Dow's constrained geometry catalysts are based on group IV transition metals that are covalently bonded to a monocyclopentadienyl group bridged with a heteroatom. The bond angle between the monocyclopentadienyl group, the titanium center and the heteroatom is less than 115°. When the alpha olefin is present in the copolymer in the range of about 10 to 20% by weight these copolymers are referred to as plastomers. When the percent alpha olefin is greater than 20% these copolymers are called elastomers. The preferred ethylene octene copolymer has the octene comonomer present in an amount less than 25%. Examples of the Dow ethylene octene copolymer have the following physical properties.

DENSITY	MOLECULAR	MELT	MELT	
<u>g/cc</u>	<u>MELT</u>	<u>INDEX</u>	<u>FLOW RATIO</u>	<u>STRENGTH</u>
	<u>WEIGHT DISTRIBUTION</u>			
Polymer 1.				
0.920	1.97	1.0	9.5	1.89

10

0.910	1.90	1.0	7.9	1.68
0.902	2.10	1.0	7.6	1.68

Molecular weight distribution is defined as the ratio of weight average molecular weight to number average molecular weight. The lower the figure, the narrower the molecular weight distribution. Melt flow ratio is defined as the ratio of melt index, as tested with a 10-kg load to the melt index with a 2-kg load. The higher the ratio, the more processable the material. Melt flow ratio is defined as melt tension measured in grams. The higher the number the greater the melt strength. Other suitable resins are the Exact resins sold by Exxon, these resins have the following characteristics:

Property	Typical properties of Exact medical grade polyethylenes					
	Value by grade					
	4028	4022	4021	4023	4024	4027
Melt index (D1238)*	10	6	22	35	3.8	4
Density, g./cc. (D-1505)	0.880	0.890	0.895	0.882	0.895	0.895
Hardness (D-2240)						
Shore A	78	84	84	80	83	89
Shore D	29	35	36	27	35	39
Tensile strength at break, p.s.i. (D-638)	2220	1700	3260	620	2840	2200
Tensile elongation at break, % (D-638)	>800	>800	>800	>800	>800	>800
Tensile impact, ft.-lb./sq. in. (D-1822)	145	130	350	280	300	340
Flexural modulus, p.s.i. (D-790)	5040	4930	3980	3100	4180	7230
Vicat softening point °F. (D-1525)	138	168	158	138	158	181

a: ASTM test method

The structure of the present invention is comprised

of an ethylene, propylene, or styrene polymer or copolymer formed by a polymerization reaction in the presence of a single site catalyst preferably a metallocene. Ethylene may be copolymerized with any suitable monomer such as  $C_3 - C_{20}$  alpha olefin including propylene butene-1, 4-methyl pentene-1, hexene-1 and octene-1. A preferred comonomer is octene-1. The preferred ethylene alpha olefin copolymer of the present invention has a density in the range of .880 gm/cc to about .920 gm/cc, a more preferred range of .890 gm/cc to about .915 gm/cc and a most preferred range of about .900 gm/cc to about .912 gm/cc.

Figure 1 shows a cross section of a three layer coextruded structure. Layer 14 is the core layer which may be a barrier layer that minimizes the transmission of oxygen through the structure. Preferred barrier materials are polyvinylidene chloride copolymers such as copolymers of vinylidene chloride and vinyl chloride or an alkyl acrylate such as methyl acrylate. Other preferred barrier material includes, ethylene vinyl alcohol, nylon or a metal foil such as aluminum. Layer 14 may also be a copolymer of ethylene and styrene formed using a single site catalyst in the polymerization reaction. The copolymer of vinylidene chloride may also be polymerized by the polymerization reaction in the presence of a single site catalyst. In addition, layer 14 may also be a polystyrene formed by a polymerization reaction in the presence of a single site catalyst. One such polystyrene is the crystalline syndiotactic polystyrene sold by Idemitsu Petro-Chemical Co., Tokyo, Japan.

*is really 1?*

On opposite sides of the core layer 14 of Figure 1 are layers 12 and 16. At least one of these layers 12 is a polymer formed by a polymerization reaction in the presence of a single site catalyst. The remaining layer 16 may be any suitable polymeric material such as a polyester, co-polyester, polyamide, polycarbonate, polypropylene, propylene-ethylene copolymer, ethylene-propylene copolymer, combinations of polypropylene and ethylene vinyl acetate copolymer, ultra low density polyethylene, low density polyethylene, medium density polyethylene, high density polyethylene, linear low density polyethylene copolymers, linear medium density polyethylene copolymer, linear high density polyethylene copolymer, ionomer, ethylene acrylic acid copolymer, ethylene ethyl acrylate copolymer, ethylene methyl acrylate copolymer, or ethylene methacrylic acid copolymer.

In an alternate embodiment, the layer 12 may be a blend of a polymer formed by a polymerization reaction in the presence of a single site catalyst and a suitable polymeric material such as is identified in connection with the description of layer 16 above.

As seen in Figure 2, the structure may also include embodiments which have a fourth layer 28 over the first layer 22 and a fifth polymeric layer 30 over the third layer 26. The composition of the fourth layer 28 may be selected from the same group of materials from which the composition of the first layer 12 or third layer 16 is selected, and the fifth layer 30 may also be the same composition as the first layer 22 or the third layer 26.

In an alternate embodiment of Figure 2, the five layer structure may have a first layer 28 similar in composition to layer 12 of Figure 1, i.e., the film may have a first layer of a polymer formed by the polymerization reaction with a single site catalyst or blends thereof with another suitable polymeric material. One or both of the second 22 and fourth 26 layers may be an adhesive layer.

The composition of adhesive layers 22 and 26 is selected for its capability to bond the core or barrier layer 24 to the surface layers 28 and 30. A variety of the well known extrudable adhesive polymers adhere well to the core or barrier layer 24. Thus, if for example layer 30 is a polypropylene, an adhesive polymer based on polypropylene is desirably selected for layer 26. Examples of such adhesives are the extrudable polymers available under the trade designations Admer QF-500, QF550, or QF-551 from Mitsui Petrochemical Company, or Exxon 5610A2.

If the composition of layer 28 or 30 is an ethylene based polymer or copolymer, an adhesive polymer based on ethylene is preferably selected for layer 22, including ethylene homopolymer and copolymers. Such a preferred adhesive composition is an ethylene vinyl acetate copolymer containing 25% to 30% by weight vinyl acetate. Other ethylene based homopolymer and copolymers, modified to enhance adhesion properties are well known under the trade names of, for example, Bynel and Plexar. Typical base polymers for these extrudable adhesives are the polyethylene and the ethylene vinyl acetate copolymers. Such adhesive polymers, including

the polypropylene-based polymers, are typically modified with carboxyl groups such as anhydride. Also acceptable as adhesives are ethylene methyl acrylate copolymers (EMA).

Additional layers may also be present in the structures of the present invention. For example, the present invention contemplates 4, 6, 7, 8, and higher numbers of layers in the film of the present invention and different combinations of layer structures may also be present. For example, there may be more than one barrier layer, i.e., two layers of polyvinylidene chloride copolymers, two layers of foil or two layers of EVOH or nylon. Alternatively, this may be a layer of EVOH and a layer of a polyvinylidene chloride copolymer or a polyamide or a polystyrene and other combinations of the core materials. The additional layers of the present invention also encompass more than one polymer formed by the polymerization reaction in the presence of a single site catalyst. The polymers may be in a layer alone or in the form of a blend. Suitable polymers for blending with an ethylene polymer formed in a polymerization reaction with a single site catalyst include other ethylene polymers formed in a polymerization reaction with a single site catalyst, LDPE, LLDPE, ULDPE, EVA, ionomers, ethylene copolymers, ethylene methyl acrylate (EMA), ethylene acrylic acid (EAA), ethyl methyl acrylic acid (EMAA), polypropylene (PP), ethylene normal butyl acrylate (ENBA), ethylene propylene copolymers (PPE). Suitable polymers for blending with a propylene polymers formed in a polymerization reaction with a single site catalyst include ethylene propylene copolymers.



Preferred blends using EVA's are those having lower VA content as they tend to yield EVA layers having better hot strength. EVA's having higher VA content tend to yield EVA layers having increased adhesion to for example, the vinylidene chloride copolymer layer. EVA's having virtually any amount of VA will have better adhesion to the vinylidene chloride copolymer layer than an ethylene homopolymer. However, good interlayer adhesion is considered desirable in the invention, and thus, steps are usually taken to enhance adhesion where no unacceptable negative effect is encountered. Thus, higher VA contents, in the range of 6% to 12% vinyl acetate are preferred, a melt index of less than 1 is also preferred. While blend amounts are shown herein in weight percent, VA contents are mole percent. Especially preferred EVA's have VA content of 7% to 9% and melt index of 0.2 to 0.8. Blends of EVA's to make up the EVA component of layers 16 and 18 are acceptable.

The structure of the present invention may be formed by any conventional process. Such processes include extrusion, coextrusion, extrusion coating, extrusion lamination, adhesive lamination and the like, and combinations of processes. The specific process or processes for making a given film which is neither oriented nor cross-linked can be selected with average skill, once the desired structure and compositions have been determined.

When the structure of the present invention is a film, the film may also be oriented either uniaxially or biaxially. Orientation can also be done by any conventional

process for forming multiple layer films. A preferred process includes the steps of coextrusion of the layers to be oriented, followed by orientation in one of the conventional processes such as blown tubular orientation or stretch orientation in the form of a continuous sheet; both being molecular orientation processes. The double bubble technique disclosure in Pahlke, U.S. Patent No. 3,456,044 is suitable for use in producing the film of this invention. The films may also be formed by a tubular water quench process. In this process the film may be extruded downwardly as a tube formed by an annular die, and carried into a water quench tank, generally with a cascade of water on the outside surface providing initial cooling. The flattened tape is withdrawn from the quench bath, is reheated (normally in a second water bath) to its orientation temperature, is stretched in the machine direction between two sets of rolls that are so rotated as to establish a linear rate differential therebetween, and is simultaneously oriented in the transverse, or cross-machine, direction as an inflated bubble trapped between the nips of the rolls. In accordance with conventional practice, the film will usually be cooled by air in the orientation zone.

The film of the present invention may also be oriented and/or cross-linked. The first step is the formation of a multiple layer film. The formation of the multiple layer film, is usually most easily accomplished by coextrusion of the desired layers. Other formation processes are acceptable so long as the resulting oriented film at the conclusion of

fabrication processing is a unitary structure.

The second step is orienting the multiple layer film. One method for accomplishing orientation is by heating the film to a temperature appropriate to molecular orientation and molecularly orienting it. The film may then be optionally heat set by holding it at an elevated temperature while its dimensions are maintained. The orientation step is preferentially carried out in line with the first step, which is the film formation step of the process.

The third step is subjecting the formed and oriented multiple layer film, to electron beam irradiation.

The amount of electron beam irradiation is adjusted, depending on the make-up of the specific film to be treated and the end use requirement. While virtually any amount of irradiation will induce some cross-linking, a minimum level of at least 1.0 megarads is usually preferred in order to achieve desired levels of enhancement of the hot strength of the film and to expand the range of temperature at which satisfactory heat seals may be formed. While treatment up to about 50 megarads can be tolerated, there is usually no need to use more than 10 megarads, so this is a preferred upper level of treatment the most preferred dosage being 2 to 5 megarads.

The third step of subjecting the film to electron beam irradiation is performed only after the multiple layer film has been formed, and after molecular orientation, in those embodiments where the film is molecularly oriented. It should be noted that, in the irradiation step, all of the

layers in the film are exposed simultaneously to the irradiation sources, such that irradiation of all the layers of the film takes place simultaneously.

In one embodiment of the process, the second step of orientation may be omitted and the unoriented multiple layer film may be cross-linked by irradiation treatment to produce a cross-linked, unoriented, multiple layer film.

### EXAMPLES

Multilayer films may be prepared according to the present invention. Biaxially stretched three layer films may be prepared by a "double bubble" process similar to that disclosed in U.S. Patent No. 3,456,044 by coextruding the following compositions through a multilayer die, biaxially stretching the coextruded primary tube. The films may also be irradiated if desired.

#### EXAMPLE 1

- Layer 1 - Copolymer of ethylene and an alpha olefin such as Hexene-1 or Octene-1 formed by the polymerization reaction in the presence of a single site catalyst or metallocene (hereinafter CEO)
- Layer 2 - Vinylidene chloride - methyl acrylate (VDC-MA) copolymer
- Layer 3 - Polyolefin. This film may be biaxially stretched and if necessary irradiated.

#### EXAMPLE 2

Layer 1 CEO  
Layer 2 VDC-MA  
Layer 3 ULDPE-EVA blend

#### EXAMPLE 3

CEO  
VDC-MA  
CEO

#### EXAMPLE 4

CEO-EVA blend  
VDC-MA  
CEO-EVA blend

#### EXAMPLE 5

LAYER 1 CEO  
LAYER 2 Nylon  
LAYER 3 CEO

#### EXAMPLE 6

CEO  
Nylon  
ULDPE-EVA

#### EXAMPLE 7

CEO-EVA blend  
Nylon  
CEO-EVA blend

EXAMPLE 8

LAYER 1 Polyolefin

LAYER 2 Styrene copolymer formed  
by the polymerization reaction  
with a single site catalyst

LAYER 3 Polyolefin

EXAMPLE 9

Polyolefin

Propylene copolymer  
formed by the  
Polymerization  
reaction with a  
single site catalyst  
PolyolefinEXAMPLE 10

LAYER 1 CEO

LAYER 2 CEO

LAYER 3 CEO

EXAMPLE 11

CEO

EVOH

ULDPE-EVA Blend

EXAMPLE 12

CEO-EVA Blend

EVOH

CEO-EVA Blend

EXAMPLE 13

LAYER 1 CEO

LAYER 2 Tie

LAYER 3 PVDC Copolymer  
or EVOH

LAYER 4 Tie

LAYER 5 ULDPE-EVA Blend

EXAMPLE 14

CEO

Tie

PVDC Copolymer  
or EVOH

Tie

CEO

EXAMPLE 15

CEO-EVA Blend

Tie

PVDC Copolymer  
or EVOH

Tie

CEO-EVA Blend

EXAMPLE 16

LAYER 1 EVA-ULDPE

LAYER 2 ULDPE or CEO

LAYER 3 PVDC Copolymer or EVOH

LAYER 4 EVA

LAYER 5 CEO or blend of CEO and EVA

The following examples may also be prepared in  
accordance with the present invention:

EXAMPLE 17

Meat Film - Forming Web  
Formed by TWQ Process  
(Tubular Water Quench Process)

LAYER 1 Nylon

LAYER 2 Tie

LAYER 3 EVOH  
LAYER 4 Tie  
LAYER 5 CEH or CEO

CEH is a copolymer of ethylene and Hexene-1 formed by the polymerization reaction in the presence of a single site catalyst or a metallocene. Other alpha olefins can be polymerized with the ethylene also.

EXAMPLE 18-20

Innerliner Films - These films can be formed either on a blown film line or by using a tubular water quench.

LAYER 1 HDPE  
LAYER 2 Blend of CEH or CEO and EVA and polybutylene

LAYER 1 HDPE  
LAYER 2 CEH or CEO and polybutylene

LAYER 1 HDPE  
LAYER 2 CEH or CEO

Example 21 and 22

Meat - Non Forming Top Web film

LAYER 1 PVDC coated PET  
LAYER 2 Adhesive (lamination)  
LAYER 3 CEO or CEH

This film may be formed by adhesive laminating a film formed of a copolymer of ethylene and an alpha olefin with the PVDC coated PET film.

LAYER 1 PVDC coated PET  
LAYER 2 LDPE - extrusion laminated  
LAYER 3 LDPE/CEH or CEO coextrusion

This film can be formed by extrusion laminating a film of PVDC

coated PET or LDPE.

Example 23

Layer 1 - Blend of two or more copolymers of ethylene and an alpha olefin polymerized in the presence of a single site catalyst or metallocene such as CEO with either CEH or CEB. CEB is a copolymer of ethylene and butene-1 formed by a polymerization reaction in the presence of a single site catalyst or a metallocene.

Example 24

Layer 1- Blend of a copolymer of ethylene and an alpha olefin formed by a polymerization reaction in the presence of a single site catalyst or a metallocene with Polyethylene or other polyolefin such as EVA, EMA, EAA, EMAA, ionomers, ENBA, PP or PPE.

The films of example 23 and 24 can either be single layer films or multi layer films where additional layers are present on layer 1.

WE CLAIM:

1. A structure comprising a layer of a polymer formed by the polymerization reaction with a single site catalyst.
2. The article of claim 1 wherein said structure is a film.
3. The film according to claim 2 wherein said layer comprises an ethylene polymer formed by the polymerization reaction with a single site catalyst.
4. The film according to claim 2 wherein said layer comprises a propylene polymer formed by the polymerization reaction with a single site catalyst.
5. The film according to claim 2 wherein said layer comprises a vinylidene chloride copolymer formed by the polymerization reaction with a single site catalyst.
6. The film according to claim 2 wherein said layer comprises a polystyrene formed by the polymerization reaction with a single site catalyst.
7. The film of claim 3 where said polymer is a copolymer of ethylene.
8. The film of claim 7 where said polymer is a copolymer of ethylene and a C<sub>3</sub>-C<sub>20</sub> alpha olefin.
9. The film according to claims 1-8 wherein said single site catalyst is a metallocene.
10. The film according to claim 8 wherein said alpha olefin is butene-1.
11. The film according to claim 8 wherein said alpha olefin is hexene-1.
12. The film according to claim 8 wherein said alpha olefin is 4-methyl pentene-1.



13. The film according to claim 8 wherein said alpha olefin is octene-1.
14. The film according to claim 8 wherein said polymeric layer is a blend of said copolymer of ethylene and an alpha olefin with a polyolefin.
15. The film according to claim 14 wherein said polyolefin is a low density polyethylene.
16. The film according to claim 14 wherein said polyolefin is a linear low density polyethylene.
17. The film according to claim 14 wherein said polyolefin is an ethylene vinyl acetate.
18. The film according to claim 14 wherein said polyolefin is a polymer of ethylene and an alpha olefin formed by the polymerization reaction with a single site catalyst.
19. The film according to claim 7-8 and 10-18 wherein said film is molecularly oriented.
20. The film according to claims 7-8 and 10-18 wherein said film is biaxially oriented.
21. The film according to claim 7-8 and 10-18 wherein said film is cross-linked.
22. The film according to claims 7-8 and 10-18 wherein said film is irradiated.
23. The film according to claims 7-8, and 10-18 further comprising a barrier layer.
24. The film according to claim 23 wherein said barrier layer is a copolymer of vinylidene chloride.
25. The film according to claim 23 wherein said barrier layer is ethylene vinyl alcohol.

26. The film according to claim 24 wherein said vinylidene chloride copolymer is a copolymer of vinylidene chloride and methyl acrylate.
27. A bag made from the film of claims 2-8, 10-18.
28. The article of claim 1 wherein said structure is a collapsible dispensing tube.
29. The tube according to claim 28 where said layer comprises an ethylene polymer formed by a polymerization reaction with a single site catalyst.
30. The tube according to claim 29 wherein said ethylene polymer is a copolymer of ethylene.
31. The tube according to claim 30 wherein said polymer is a copolymer of ethylene and  $C_3$ - $C_{20}$  alpha olefin.
32. The tube according to claim 31 wherein said alpha olefin is octene-1.
33. The article of claim 1 where said structure is a pouch.
34. The article of claim 1 wherein said structure is a container.
35. A structure comprising a layer of polymer formed by the polymerization reaction with a metallocene catalyst system.
36. The article of claim 35 wherein said structure is a film.
37. The film according to claim 36 wherein said layer comprises an ethylene polymer formed by the polymerization reaction with a metallocene catalyst system.
38. The film according to claim 36 wherein said layer comprises a propylene polymer formed by the

polymerization reaction with a metallocene catalyst system.

39. The film according to claim 36 wherein said layer comprises vinylidene chloride copolymer formed by the polymerization reaction with a metallocene catalyst system.
40. The film according to claim 36 wherein said layer comprises a polystyrene formed by the polymerization reaction with a metallocene catalyst system.
41. The film of claim 37 where said polymer is a copolymer of ethylene.
42. The film of claim 41 where said polymer is a copolymer of ethylene and a C<sub>3</sub>-C<sub>20</sub> alpha olefin.
43. The film according to claim 42 wherein said alpha olefin is butene-1.
44. The film according to claim 42 wherein said alpha olefin is hexene-1.
45. The film according to claim 42 wherein said alpha olefin is 4-methyl pentene-1.
46. The film according to claim 42 wherein said alpha olefin is octene-1.
47. The film according to claim 42 wherein said polymeric layer is a blend of said copolymer of ethylene and an alpha olefin with a polyolefin.
48. The film according to claim 47 wherein said polyolefin is a low density polyethylene.
49. The film according to claim 47 wherein said polyolefin is a linear low density polyethylene.

50. The film according to claim 47 wherein said polyolefin is an ethylene vinyl acetate.
51. The film according to claim 47 wherein said polyolefin is a polymer of ethylene and an alpha olefin formed by the polymerization reaction with a metallocene catalyst system.
52. The film according to claim 40-41 and 42-51 wherein said film is molecularly oriented.
53. The film according to claims 40-41 and 42-51 wherein said film is biaxially oriented.
54. The film according to claims 40-41 and 42-51 wherein said film is cross-linked.
55. The film according to claims 40-41 and 42-51 wherein said film is irradiated.
56. The film according to claims 40-41 and 42-51 further comprising a barrier layer.
57. The film according to claim 56 wherein said barrier layer is a copolymer of vinylidene chloride.
58. The film according to claim 56 wherein said barrier layer is ethylene vinyl alcohol.
59. The film according to claim 57 wherein said vinylidene chloride copolymer is a copolymer of vinylidene chloride and methyl acrylate.
60. A bag made from the film of claims 35-53.
61. The article of claim 35 wherein said structure is a collapsible dispensing tube.
62. The tube according to claim 61 where said layer comprises an ethylene polymer formed by a polymerization reaction

with a metallocene catalyst system.

63. The tube according to claim 62 wherein said ethylene polymer is a copolymer of ethylene.
64. The tube according to claim 63 wherein said polymer is a copolymer of ethylene and  $C_3$ - $C_{20}$  alpha olefin.
65. The tube according to claim 64 wherein said alpha olefin is octene-1.
66. The article of claim 35 where said structure is a pouch.
67. The article of claim 35 wherein said structure is a container.
68. The article of claim 1 wherein said layer comprises an amide polymer formed by the polymerization reaction with a single site catalyst.
69. The article claim 35 where said layer comprises an amide polymer formed by the polymerization reaction with a metallocene catalyst system.
70. The article of claims 68 or 69 wherein said amide is a nylon.
71. The film according to claim 2 further comprising a layer of HDPE on said layer of a polymer formed by the polymerization reaction with a single site catalyst.
72. The film according to claim 71 where said polymer is a copolymer of ethylene and an alpha olefin.
73. The film according to claim 72 wherein said alpha olefin is hexene-1.
74. The film according to claim 72 wherein said alpha olefin is octene-1.
75. The film according to claim 73 or 74 wherein said polymer

- layer further includes a polybutylene.
76. The film according to claim 2 further comprising a layer of polyethylene terephthalate (PET) coated with a polyvinylidene chloride copolymer and an adhesive layer adhering said PET layer with said polymer layer.
  77. The film according to claim 76 wherein said adhesive is low density polyethylene.
  78. The film according to claim 76 wherein said film is formed by adhesive lamination.
  79. The film according to claim 77 wherein said film is formed by extrusion lamination.
  80. The film according to claim 2 wherein said film further comprises a second layer of an adhesive, a third or core layer of ethylene vinyl alcohol, a fourth layer of an adhesive and a fifth layer of nylon.
  81. The film of claim 80 formed by means of the tubular water quench process.
  82. The film according to claim 75 wherein said polymer layer further includes ethylene vinyl acetate.
  83. A process for forming a structure comprising the steps of providing a layer of a polymer formed by the polymerization reaction with a single site catalyst.
  84. The process according to claim 83 wherein said structure is a film.
  85. The process according to claim 84 wherein said film comprises a layer of an ethylene polymer formed by the polymerization reaction with a single site catalyst.
  86. The process according to claim 84 wherein said film

comprises a layer of an ethylene polymer formed by the polymerization reaction with a metallocene catalyst system.

87. The process according to claims 85 or 86 wherein said film is molecularly oriented.
88. The process according to claim 87 wherein said film is biaxially oriented.
89. The process according to claim 88 wherein said film is irradiated.
90. The process according to claim 87 wherein said polymer is a copolymer of ethylene and an alpha olefin.
91. The film according to claim 24 wherein said film is biaxially oriented and irradiated.
92. The film according to claim 25 wherein said film is biaxially oriented and irradiated.
93. The film according to claim 26 wherein said film is biaxially oriented and irradiated.
94. The film according to claim 57 wherein said film is biaxially oriented and irradiated.
95. The film according to claim 58 wherein said film is biaxially oriented and irradiated.
96. The film according to claim 59 wherein said film is biaxially oriented and irradiated.

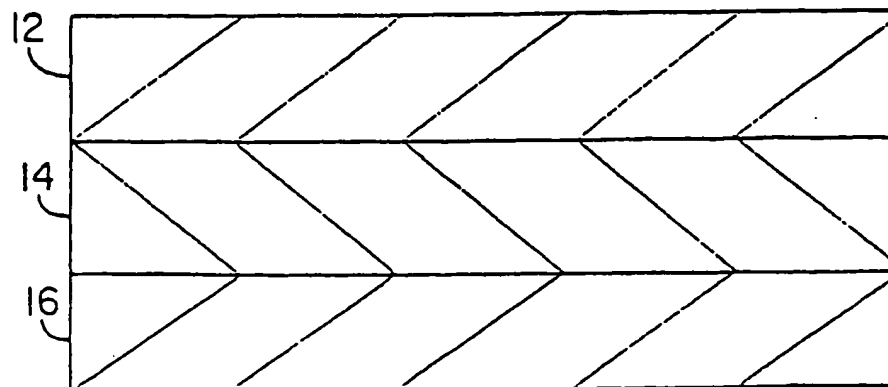


FIG. 1

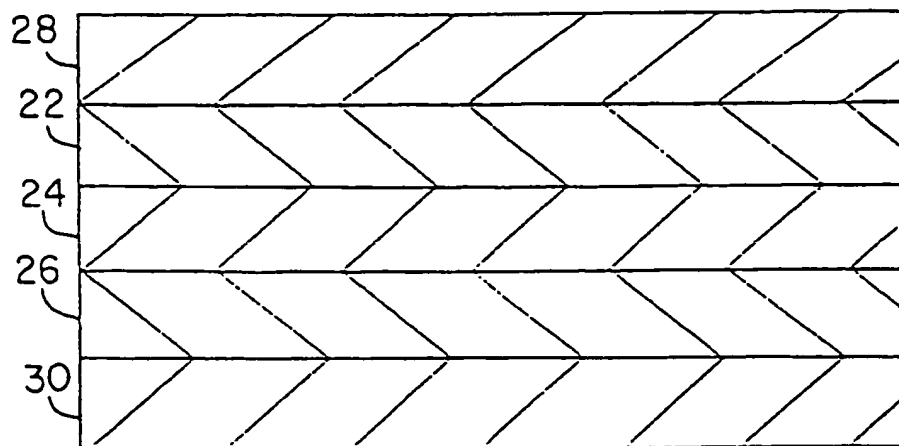
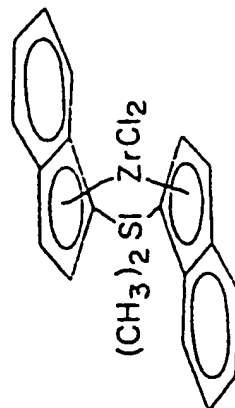
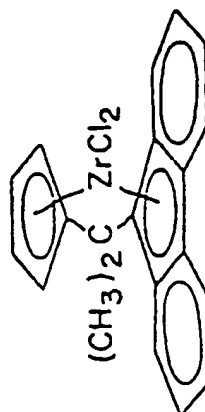
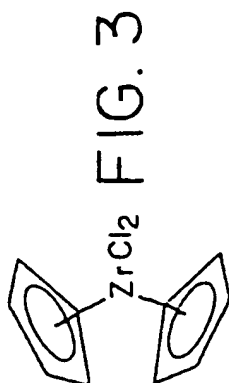
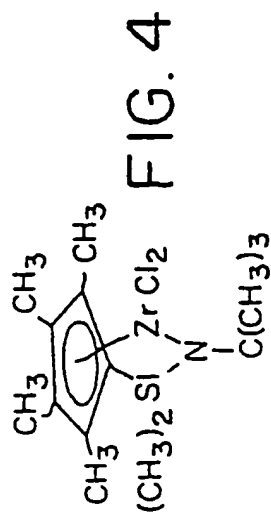


FIG. 2

SUBSTITUTE SHEET (RULE 26)





<b>A. CLASSIFICATION OF SUBJECT MATTER</b> IPC(5) : B32B 31/00 US CL : 428/474.4, 500 According to International Patent Classification (IPC) or to both national classification and IPC		
<b>B. FIELDS SEARCHED</b> Minimum documentation searched (classification system followed by classification symbols) U.S. : 428/474.4, 500		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched NONE		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) NONE		
<b>C. DOCUMENTS CONSIDERED TO BE RELEVANT</b>		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US, A, 4,871,705 (HOEL) 03 October 1989, column 2, lines 33-42. <i>Process for Production of high molecular weight EAO elastomer - metallocene aluminum catalyst.</i>	1-8,10-18, 28-51 and 61-96
Y	ENCYCLOPEDIA OF POLYMER SCIENCE AND ENGINEERING, 1987, Volume 7, 'FILMS, MANUFACTURE' pages 88-89.	1-8,10-18 28-51 and 61-96
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" documents defining the general state of the art which is not considered to be part of particular relevance "E" earlier documents published on or after the international filing date "L" documents which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reasons (as specified) "O" documents referring to an oral disclosure, use, exhibition or other means "P" documents published prior to the international filing date but later than the priority date claimed "T" later documents published after the international filing date or priority date and not in conflict with the application but cited to understate the principle or theory underlying the invention "X" documents of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" documents of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "A" documents members of the same patent family		
Date of the actual completion of the international search 29 AUGUST 1994		Date of mailing of the international search report SEP 27 1994
Name and mailing address of the ISA/US Commissioner of Patents and Trademarks Box PCT Washington, D.C. 20231 Facsimile No. (703) 305-3230		Authorized officer <i>Gene Leard for</i> E.ROLLINS BUFFALOW Telephone No. (703) 305-2351